



Managing Water Quality for Public Health

Department of Agriculture, Forest and Food Sciences University of Torino

October 14th 2015

Autonomous system design for continuous monitoring of metals in water

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<http://www.polito.it/micronanotech>

<http://www.chilab.polito.it>

Prof. Fabrizio Pirri

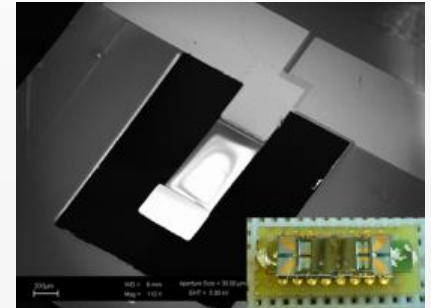
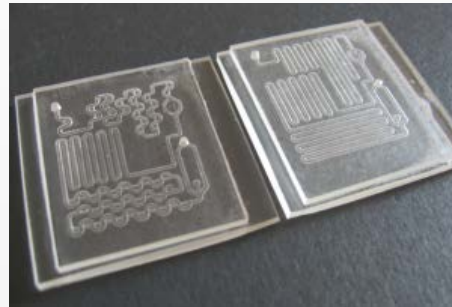
Research Leader

Staff

- 5 Professors
- 8 Permanent Researchers
- 16 Fellowships / Post Doc
- 8 PhD students
- 3 Technicians
- 2 Administratives

Mission

- fundamental research on materials and processes for micro- and nano-technologies
- design and fabrication of MEMS and nanostructures
- technological transfer
- education



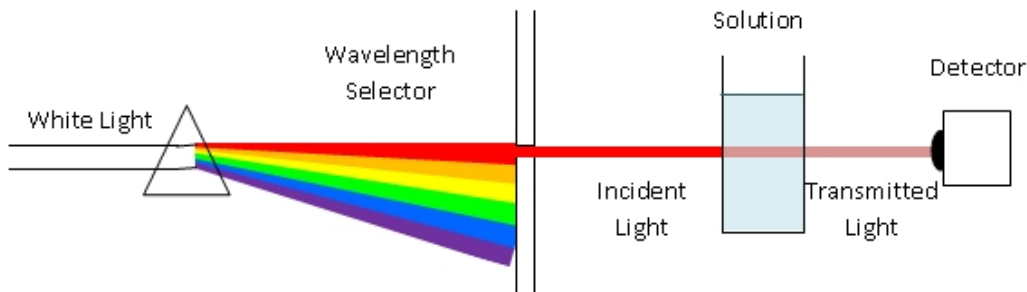
1. Spectrophotometry techniques
2. Calibration curves Spectrophotometry technique
3. Design and engineering of a portable instrument
4. Waste water monitoring
5. Fresh water monitoring
6. AUV payload engineering
7. Results and Future developments

Several techniques such as X-ray fluorescence, atomic fluorescence spectrometry, chromatography, atomic absorption spectrometry, etc. have been used for the simultaneous determination of different ions in different samples.

Among the most widely used analytical methods are those based on the **UV-Vis spectrophotometry techniques**, due to the resulting *experimental rapidity, simplicity* and the *wide application*.

Nowadays quantitative spectrophotometry has been greatly improved by the use of a variety of **multivariate statistical method**; particularly principle component regression (PCR) and **partial least squares regression (PLS)**. PLS regression has been found important in handling regression tasks in case there are many variables. **PLS allows to simultaneous determination of different ions in water compounds.**

Spectrophotometry is a method to measure how much a chemical substance absorbs light by measuring the intensity of light as a beam of light passes through sample solution. The basic principle is that each compound absorbs or transmits light over a certain range of wavelength.

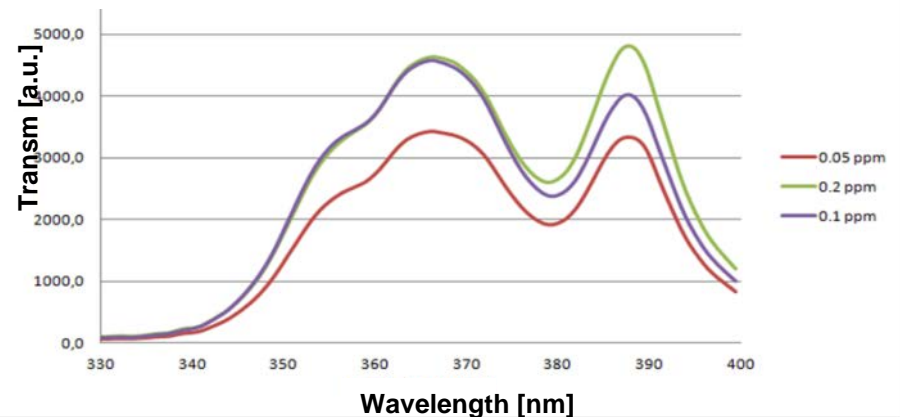


The difference between the incident and transmitted light indicates the absorbance

$$\log I_0/I = \text{Absorbance}$$
$$\text{Beer's Law}$$
$$\text{Absorbance} = \epsilon \times \text{conc} \times l$$

The adsorption spectra of three different Cr concentrations in drinking water has been compared. The finger print is obtained by subtracting the reference spectrum.

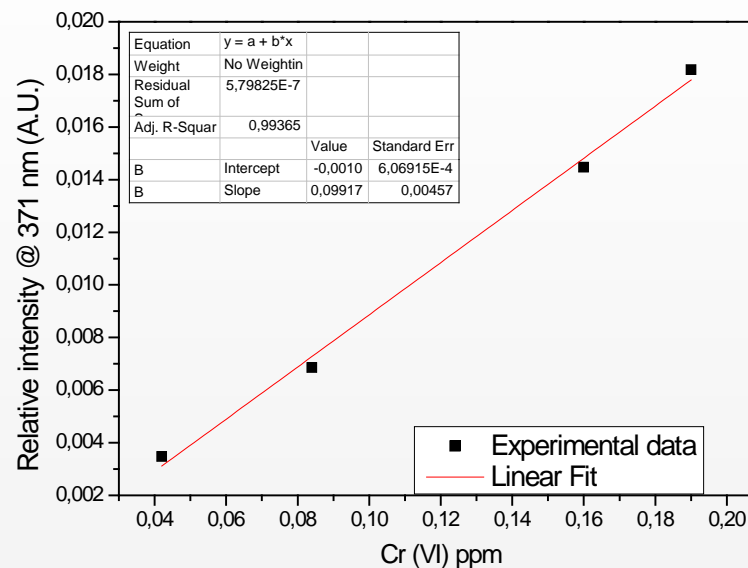
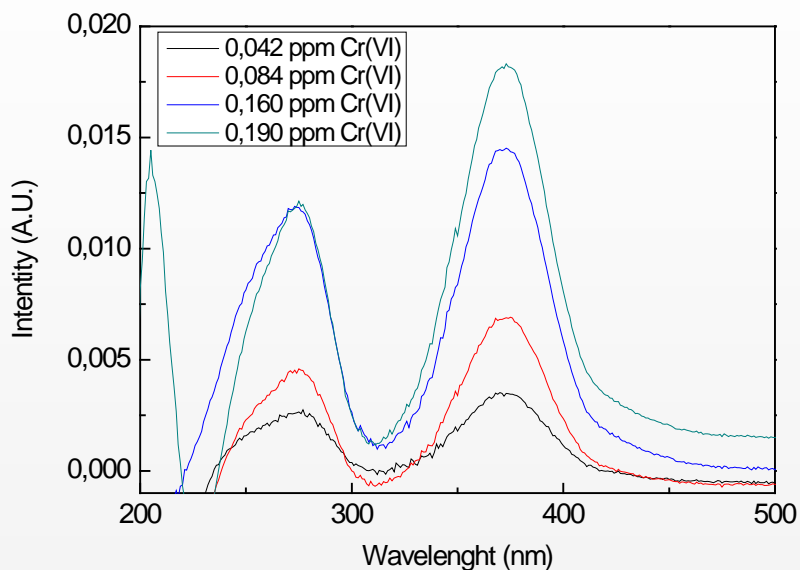
Cr spectra



Calibration curves Spectrophotometry technique

Spectrophotometry technique applied for the measurement of Cr (VI) concentration in waste water

The quantitative analysis is obtained by considering the absorption peak at 371 nm.

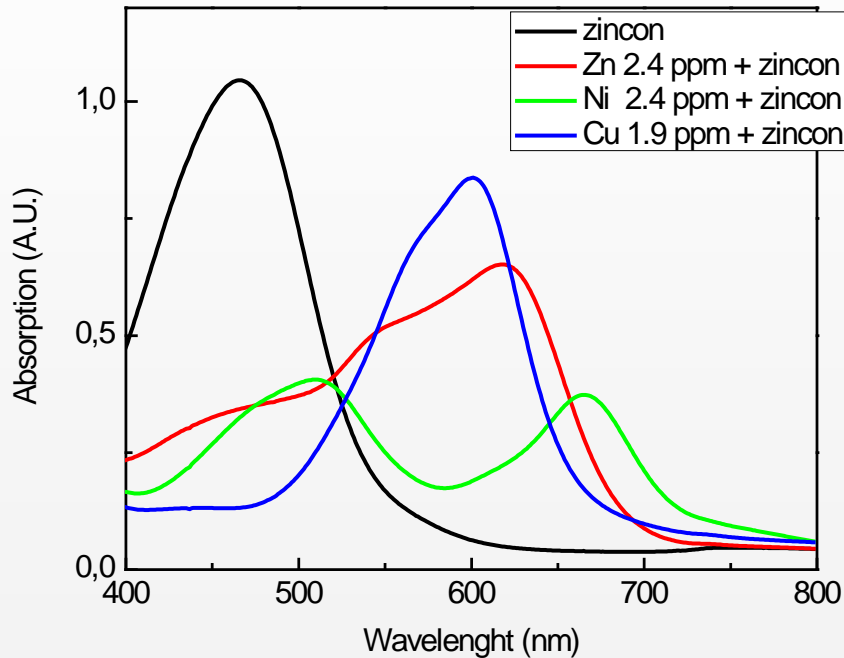


Quantitative determination of hexavalent chromium in aqueous solutions by UV-Vis spectrophotometer

M. C. Fournier-Salauun and P. Salauun

Central European Journal of Chemistry 5(4) 2007 1084–1093

The partial least squares (PLS) has been applied to the simultaneous determination of the divalent ions of **copper**, **nickel** and **zinc** based on the formation of their complexes with 2-carboxy-2-hydroxy-5-sulfoformazyl benzene (zincon). The absorption spectra were recorded in the visible spectrum (400 - 800 nm).



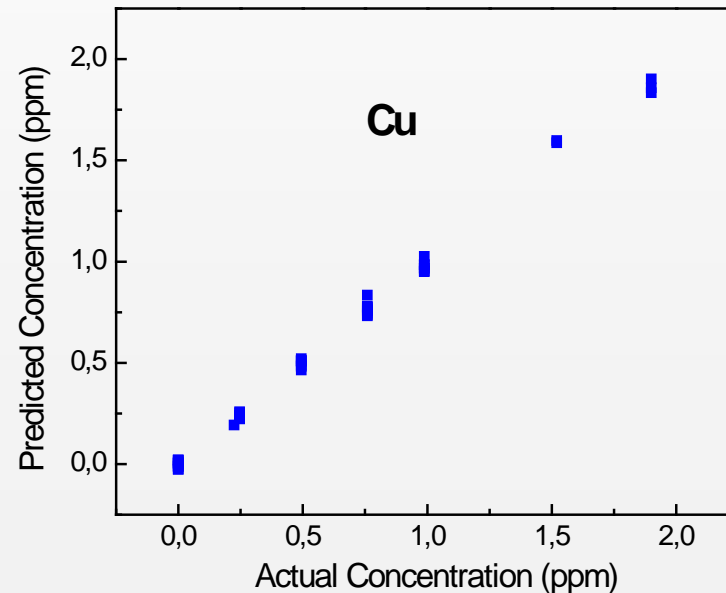
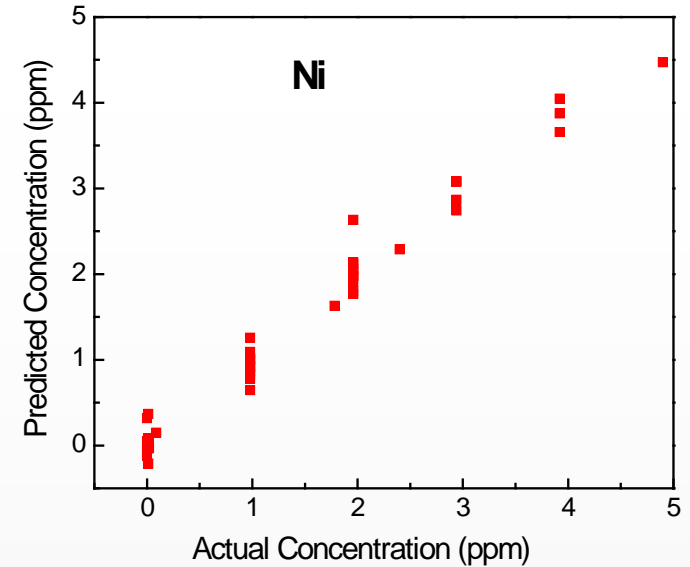
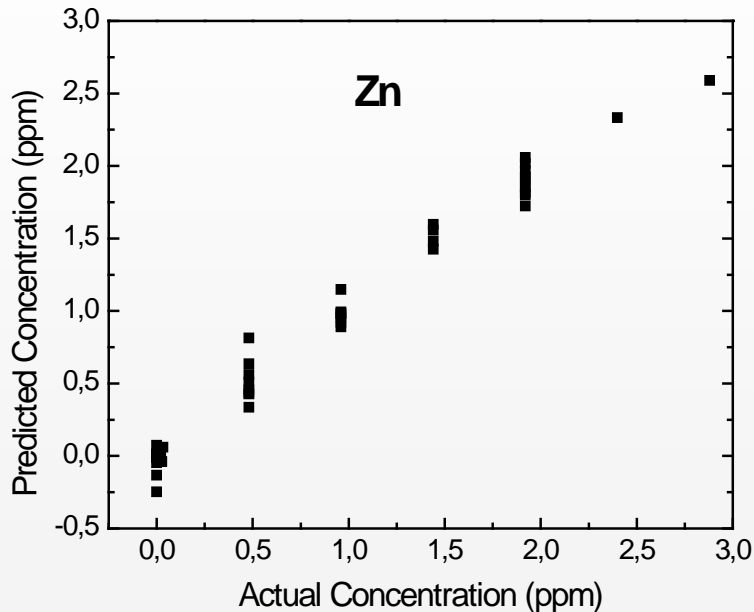
Spectra of different solution are different, nevertheless the superposition presence does not permit a direct measurement by means of absorption spectra.



PLSR: partial least squares regression is required

Calibration curves Spectrophotometry technique

Experimental results: Plots of predicted concentration vs. actual concentration for three cations (Copper, Nickel, Zinc) in the prediction set.

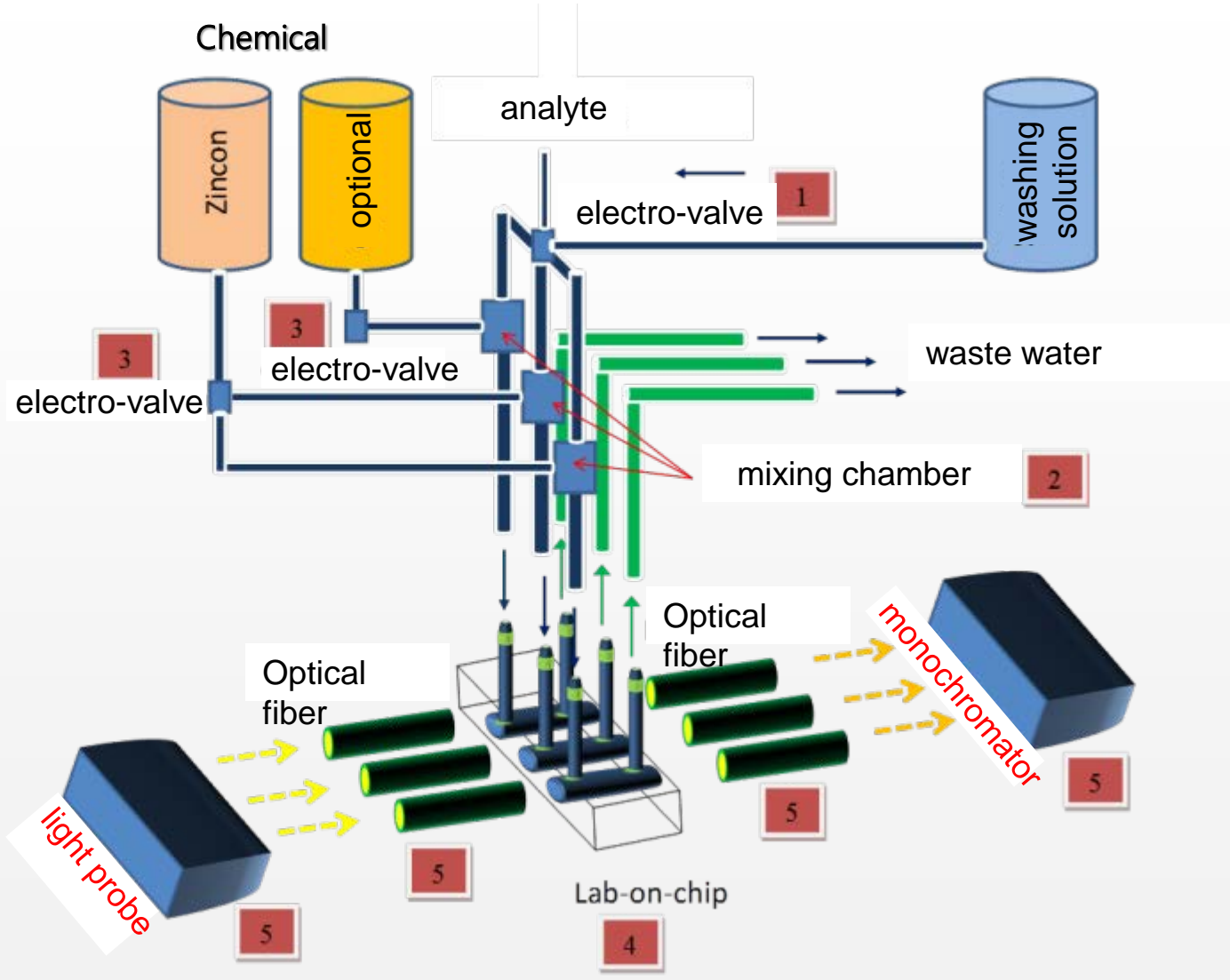
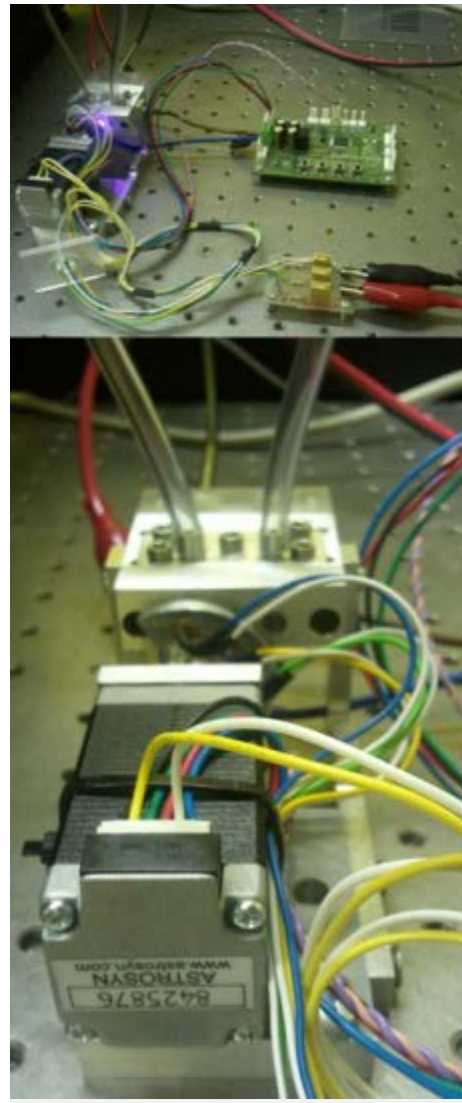


Calibration curves Spectrophotometry technique

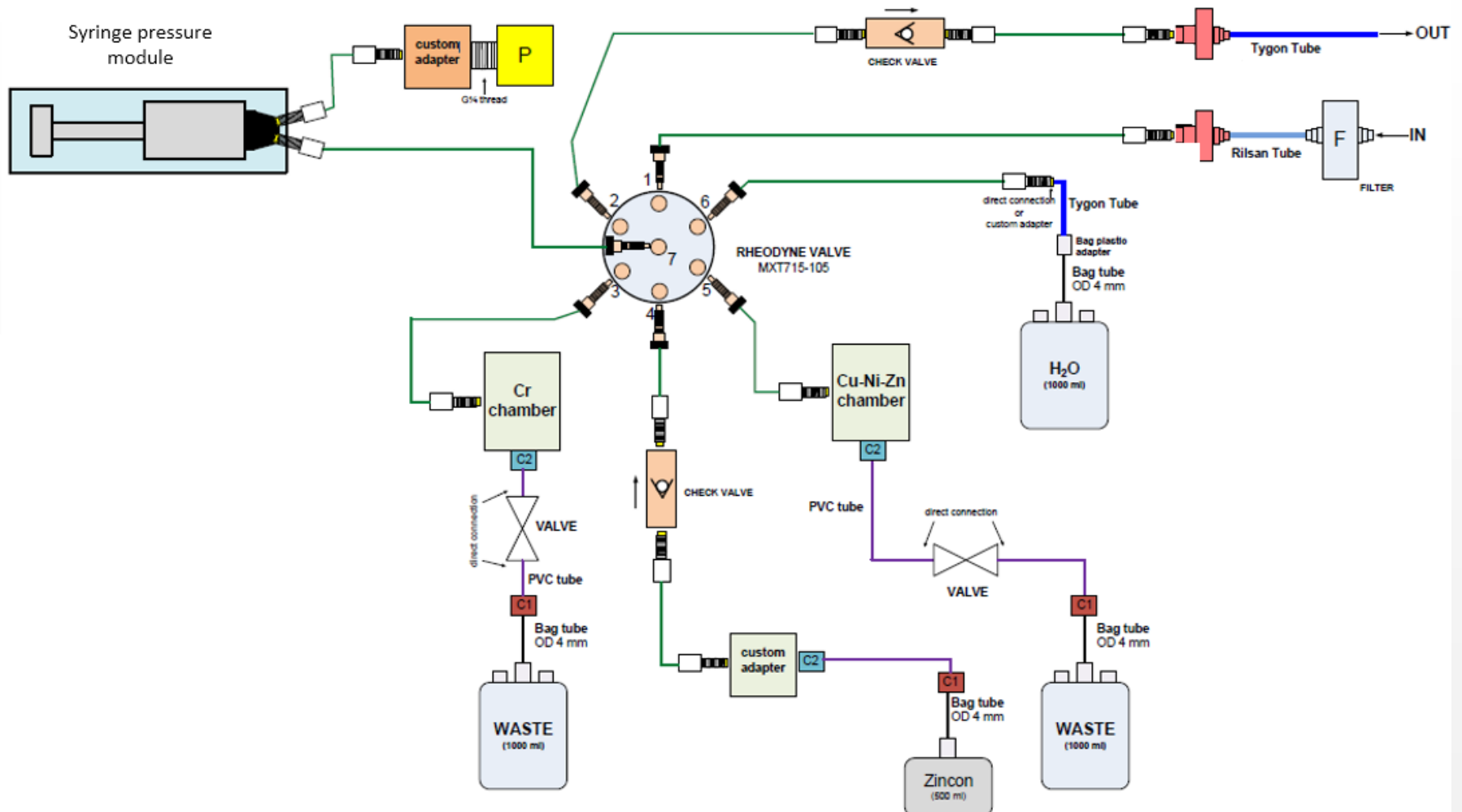
In order to test the reliability of the proposed method different tests has been performed in a variety of synthetic solutions. The results of the prediction are summarized in the table.

Sample	Zn (ppm)	Zn prediction (ppm)	Ni (ppm)	Ni prediction (ppm)	Cu (ppm)	Cu prediction (ppm)
Z5	0,25	0,22	2,9	3	0,96	0,98
Z25	0,76	0,75	1,9	1,8	1,9	1,8
Z36	1,9	1,8	0,9	0,7	1	1
12906-0,45um	0	0,01	0,01	0	0,02	0,05
12906-8ul	0	0,01	0,01	-0,2	0,02	0
12906-100ul	0	0,02	0,01	0,3	0,01	-0,01

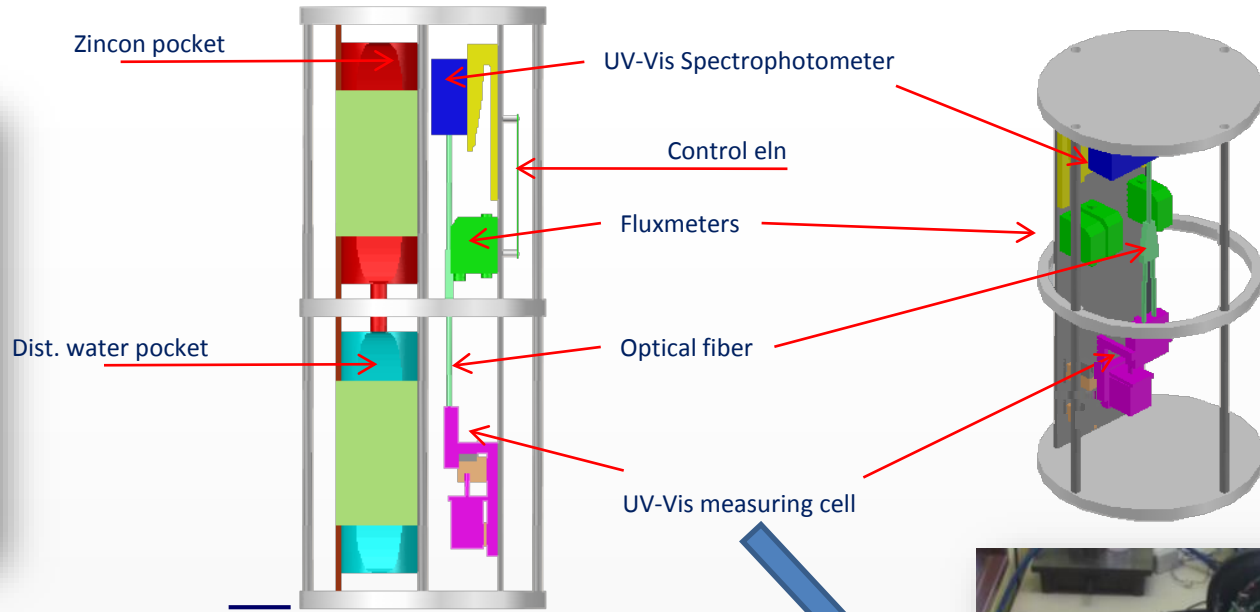
Functional schema and preliminary tests



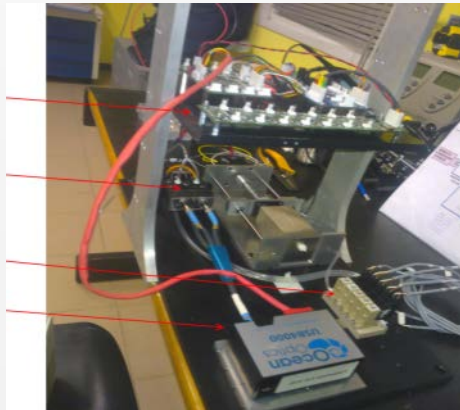
Assembly schema



Design and engineering of a portable instrument



SMAT S.p.A.



Eni S.p.A.



Waste water monitoring test – Castiglione T.se (TO) - Italy



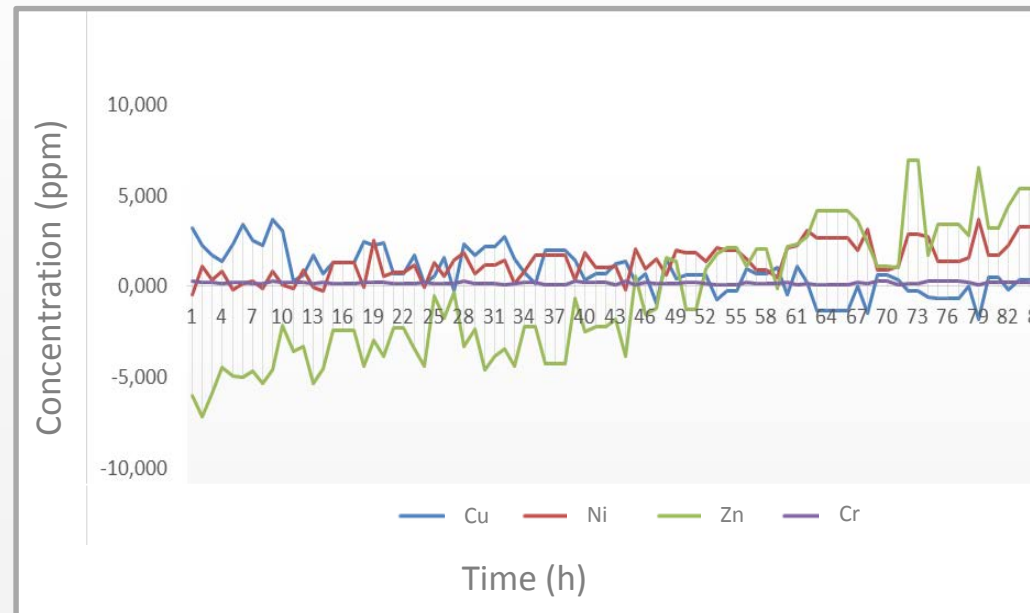
Ni, resolution	0.8	ppm
Cu, resolution	0.8	ppm
Cr VI, resolution	0.8	ppm
Zn, resolution	0.8	ppm
Measurement range	0 ÷ 10	ppm



WMA laboratory test



WMA installed



Continuous autonomous monitoring – Preliminary data

Fresh water monitoring test – Torino - Italy



WMA installed



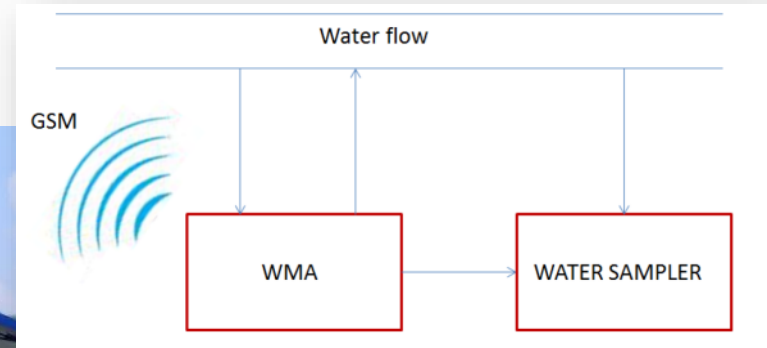
Fresh water SMAT's
delivery point



- Test performed thanks to the cooperation with SMAT S.p.A. and the grant POR-FESR 2007/2013 by Piedmont Region

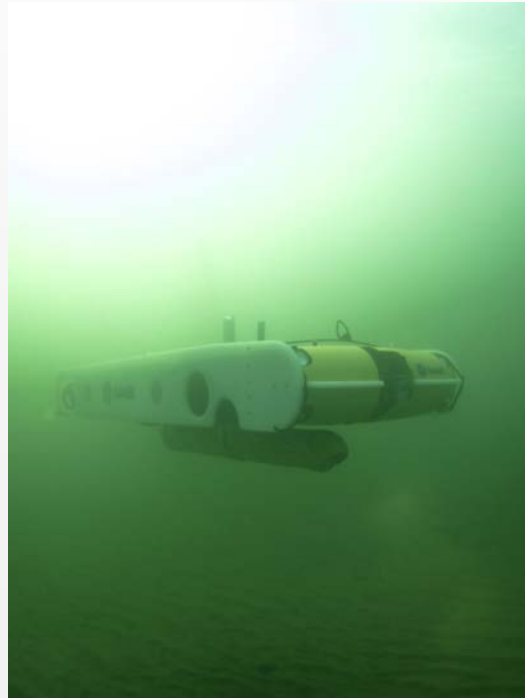


- Continuous autonomous monitoring – Preliminary data
- Data collected by GSM



Measurement data sending

AUV payload engineering - test on VATTERN Lake, Motala -Sweden



AUV payload engineering - test on VATTERN Lake, Motala -Sweden

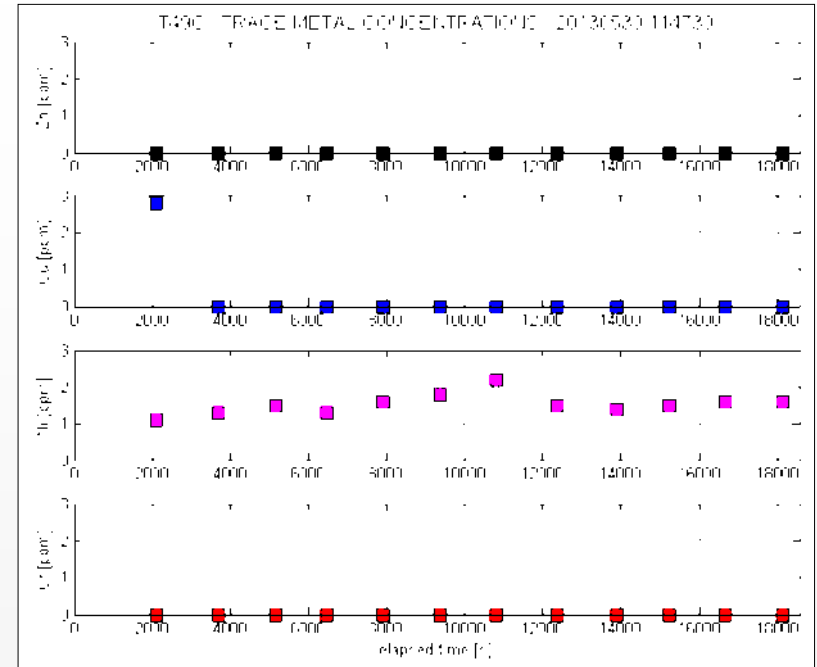
- E-POD#3 successfully operated
- Spectrophotometric technique demonstrated
- All functionalities verified, no failure occurred

- Bias on Ni observed during field tests, not present during calibration

- Laboratory analysis on water samples collected during the tests
 - confirmed in-situ measurements of Cu, Zn, Cr
 - anomaly of in-situ measure of Ni possibly due to interference of Zincon with organic substances (absorption in the same wavelength)

- Expansion to other parameters (further trace metals, hydrocarbons) possible

- method sensitivity presently 0.7 ppm, optimization required to reach 100 ppb target (best lab.technique)

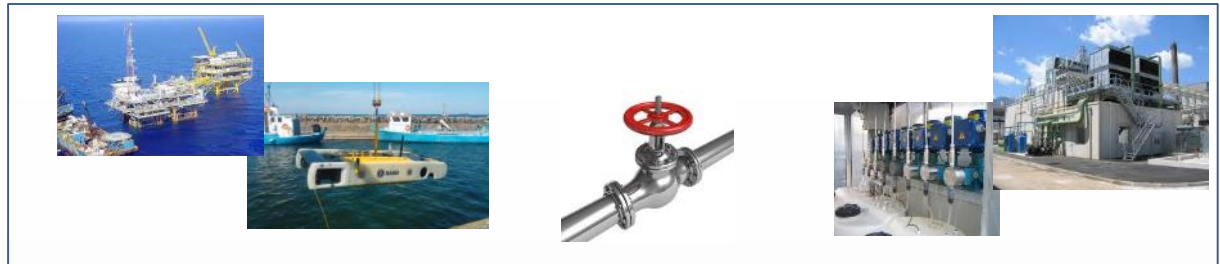


- Prototype analyser developed, **laboratory calibrated** and **fields tested**
- Concept validated (TRL 2), **optimisation work required**
- A possible qualification program (target TRL 4) should include
 - System design review; study possible optimisation of the system layout and components (and opportunity to add new parameters)
 - Laboratory tests to optimise analytical cycles and explore the limits of the system (sensitivity, repeatability, long-term duration, pressure, temperature, long-term etc.)
 - Test in a real environment (e.g. characterised by industrial pollution)

Thank you for the attention

Particular thanks to:

- Paola Rivolo, PhD
- Alessandro Virga, PhD



Thanks to:

SMAT S.p.A.

Eni S.p.A.

Microla Optoelectronics

S.r.l. for the engineering of the system.

